



# **Geobarrettin D, a Rare Herbipoline-Containing 6-Bromoindole Alkaloid from** *Geodia barretti*

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**Abstract:** Geobarrettin D (1), a new bromoindole alkaloid, was isolated from the marine sponge *Geodia barretti* collected from Icelandic waters. Its structure was elucidated by 1D, and 2D NMR (including <sup>1</sup>H-<sup>15</sup>N HSQC, <sup>1</sup>H-<sup>15</sup>N HMBC spectra), as well as HRESIMS data. Geobarrettin D (1) is a new 6-bromoindole featuring an unusual purinium herbipoline moiety. Geobarrettin D (1) decreased secretion of the pro-inflammatory cytokine IL-12p40 by human monocyte derived dendritic cells, without affecting secretion of the anti-inflammatory cytokine IL-10. Thus, compound 1 shows anti-inflammatory activity.

Keywords: 6-bromoindole; herbipoline; Geodia barretti; anti-inflammatory activity; dendritic cells

# 1. Introduction

The indole nucleus is an important element of many natural and synthetic molecules possessing significant biological activities. Indole has been termed a "privileged structure" in drug discovery and a common starting point for drug development or lead optimization [1,2]. Marine indole alkaloids comprise a large and complex class of natural products; most marinederived indole metabolites are halogenated by bromine [2–4]. The presence of halogen substituents on the indole ring profoundly influences biological activity [2,3]. Interestingly, the Br substituent generally resides at C-5, less commonly at C-6, or at both C-5 and C-6 [4]. Additional modifications of the bromoindole core include C-substitutions by O, C (often prenyl) and N groups, pyrimidyl, 2-aminopyrimidyl groups or more complex polycyclic ring systems [5–11]. Bromoindoles have been reported to have anti-inflammatory, antibacterial, antifungal, antitumor, antioxidant, antifouling, and antiplasmodial activities [1,3,4,12,13].

The marine sponge *Geodia barretti* is the source of bromoindole alkaloids [14–18] and several *N*-methylated nucleosides [19]. In our previous study, three new 6-bromoindole derivatives were isolated from *G. barretti* collected in Icelandic waters [20]; of these, geobarrettins B and C exhibited anti-inflammatory activity [20]. As part of our ongoing investigation to find novel anti-inflammatory compounds, we report, here, the bromoindole geobarrettin D (1) (Figure 1, as the TFA salt) with potential anti-inflammatory effect measured by decreased pro-inflammatory cytokine secretion of human monocyte-derived dendritic cells (DCs).



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Figure 1. Structure of geobarrettin D (1).

## 2. Results

The lyophilized sponge *G. barretti* was extracted with  $CH_2Cl_2/MeOH$  (v/v 1:1). After removal of solvent, the resulting crude extract was resuspended in MeOH/H<sub>2</sub>O (9:1) and solvent-partitioned into five fractions of increasing polarity (hexane, CHCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, *n*-BuOH, and H<sub>2</sub>O) using a modified Kupchan method [21,22]. The CHCl<sub>3</sub> and CH<sub>2</sub>Cl<sub>2</sub> fractions were combined and purified by RP C18 HPLC to afford the 6-bromoindole derivative, geobarrettin D (1, Figure 1).

# 2.1. Structural Elucidation

The HRMS of geobarrettin D (1) exhibited molecular ion isotopomers m/z 458.1343/ 460.1316 ([M]+) in a 1:1 ratio, indicating the presence of one Br and a molecular formula  $C_{20}H_{25}^{79}BrN_7O$ , corresponding to 12 degrees of unsaturation. The <sup>13</sup>C NMR data (Table 1, see Supplementary Materials) showed 18 signals which were matched to the C content of the molecular formula, including five methyls ( $\delta_C$  54.8 (×3), 36.2, and 32.1), one sp<sup>3</sup> methylene ( $\delta_C$  69.3), one aliphatic methine ( $\delta_C$  45.4), five aromatic methines ( $\delta_C$  140.1, 125.9, 124.2, 120.8, and 115.9), four quaternary aromatic carbons ( $\delta_{\rm C}$  139.1, 125.2, 117.0, and 113.4), four quaternary heteroatom-bonded sp<sup>2</sup> carbons ( $\delta_{\rm C}$  155.0, 154.9, 151.0, and 110.0) (Table 1). The most intense MS peak at m/z 399.0603/401.0585 ([M-N(CH<sub>3</sub>)<sub>3</sub>]<sup>+</sup>) (Figure S10) was derived from a neutral loss of trimethylamine  $(-N(CH_3)_3)$ . IR bands (3254, 1182, and  $1131 \text{ cm}^{-1}$ ) implied the presence of OH and/or NH functionalities. The <sup>1</sup>H NMR data of **1** in  $CD_3OD$  (Table 1) exhibited signals due to five aromatic protons, five *N*-methyl groups, [δ<sub>H</sub> 4.11(3H, s, 11-Me), 3.91(3H, s, 12-Me), and 3.27 (9H, s, 1"-NMe)], a methine proton, and a methylene group. Three aromatic signals at  $\delta_{\rm H}$  7.65 (1H, d, J = 8.5 Hz, H-4'), 7.60 (1H, d, *J* = 1.6 Hz, H-7') and 7.24 (1H, dd, *J* = 8.5, 1.6 Hz, H-5') indicated the presence of a 1,2,4-trisubstituted benzene ring. The <sup>1</sup>H NMR spectrum, recorded in  $D_2O/H_2O$  (1:9), showed the presence of a downfield exchangeable proton ( $\delta_{\rm H}$  10.60), which is diagnostic of an NH proton in an indole ring, and confirmed by a weak coupling (J = 2.0 Hz) to the isolated aromatic proton at  $\delta_{\rm H}$  7.53 in the pyrrole ring. Correlations observed in the HMBC spectrum (Figure 2) allowed the definement of the substitution pattern and NMR assignments of the indole: H-4' to C-3', C-6', and C-7a', from H-5' to C-3a' and C-7', and from H-7' to C-5' and C-3a'. H-2' also showed a correlation to C-3', C-3a', and C-7a'. The NH exchangeable proton H-1' showed correlations to C-2', C-3', C-3a', and C-7a'. The indole assignment was supported by the presence of several bands in the UV-vis spectrum  $(\lambda_{\text{max}} 228, 261, \text{ and } 287 \text{ nm}).$ 

Br-substitution at C-6' was deduced by a comparison of the chemical shifts of the aromatic carbons of related 6-bromoindole alkaloids [16,18,23]. Thus, geobarrettin D (1) was defined as a 3-substituted 6-bromoindole alkaloid. The <sup>1</sup>H-<sup>1</sup>H COSY correlations from H-3" ( $\delta_{\rm H}$  6.05 (1H, t, J = 6.3 Hz)) to H-2" ( $\delta_{\rm H}$  4.07 (1H, dd, J = 13.7, 5.9 Hz); 4.15 (1H, dd, J = 13.7, 6.8 Hz)) and HMBC correlations of (CH<sub>3</sub>)<sub>3</sub>-N ( $\delta_{\rm H}$  3.30 (9H, s))/C-2" ( $\delta_{\rm C}$  69.3), H-2"/C-3" ( $\delta_{\rm C}$  45.4) and H-3"/C-2" indicated the presence of 2,2-disubstituted *N*,*N*,*N*-trimethylethanaminium group; further support of the connectivity of 6-bromo-indol-

3-yl moiety and the *N*,*N*,*N*-trimethylethanaminium group were provided by additional HMBC correlations: H-3"/C-3' ( $\delta_C$  113.4), H-3"/C-2' ( $\delta_C$  125.9), H-3"/C-3a' ( $\delta_C$  125.2), and H-2"/C-3' (Figure 2).



Figure 2. Key <sup>1</sup>H-<sup>13</sup>C HMBC and <sup>1</sup>H-<sup>15</sup>N HMBC correlations for compound 1.

The balance of the molecular formula C<sub>20</sub>H<sub>25</sub><sup>79</sup>BrN<sub>7</sub>O of geobarrettin D (1), C<sub>7</sub>N<sub>5</sub>H<sub>8</sub>O, after accounting for the 6-bromoindole and 2,2-disubstituted *N*,*N*,*N*-trimethylethanaminium moieties, required another six degrees of unsaturation. The HMBC cross-peak H-3"/C-2 ( $\delta_{\rm C}$  154.9) revealed that the C<sub>7</sub>N<sub>5</sub>H<sub>8</sub>O unit was connected to C-3" through a C-*N* bond, which explains the downfield chemical shift of C-3" ( $\delta_{\rm C}$  45.4). Analysis of the <sup>13</sup>C NMR data revealed the seven remaining carbons as non-protonated sp<sup>2</sup> carbons with chemical shifts of  $\delta_{\rm C}$  155.0, 154.9, 151.0, 140.1, 110.0 and two sp<sup>3</sup> carbons  $\delta_{\rm C}$  36.2, 32.1 (Table 1). The <sup>1</sup>H NMR chemical shift of the non-exchangeable  $\delta_{\rm H}$  9.01 lacked an expected cross-peak in the HSQC spectrum, but strong symmetric 'satellite peaks' appearing in the HMBC, centered on  $\delta_{\rm C}$  140.1, were due to <sup>1</sup>*J*<sub>CH</sub> 'breakthrough' (<sup>1</sup>*J*<sub>H8-C8</sub> = 220 Hz) [24,25]: the large magnitude is consistent with a five-membered heterocycle [26]. Long-range correlations were also seen from H-8 [ $\delta_{\rm H}$  8.81 (1H, s, H-8)] to two *N*-methyl groups ( $\delta_{\rm C}$  36.2 and 32.1 ppm) in addition to C-4 and C-5 ( $\delta_{\rm C}$  151.0 and 110.0, respectively). These data are reconciled by an *N*,*N*-dimethyl imidazole ring.

H-detected <sup>15</sup>N-heteronuclear 2D NMR experiments (<sup>1</sup>H–<sup>15</sup>N HSQC and <sup>1</sup>H-<sup>15</sup>N HMBC in D<sub>2</sub>O/H<sub>2</sub>O, 1:9) were also recorded. The correlations from H-8 [ $\delta_{\rm H}$  8.81 (1H, s, H-8)] to N-9 ( $\delta_{\rm N}$  157.3), N-7 ( $\delta_{\rm N}$  156.4), from CH<sub>3</sub>-12 ( $\delta_{\rm H}$  3.91 (3H, s)) to N-9, from CH<sub>3</sub>-11 ( $\delta_{\rm H}$  4.11 (3H, s)) to N-7 in <sup>1</sup>H–<sup>15</sup>N HMBC and <sup>1</sup>H–<sup>13</sup>C HMBC spectra further supported a *N*,*N*-dimethyl imidazolinium ring. The latter partial structures, together with the last two degrees of unsaturation, were assembled with the remaining quaternary C and three N atoms to complete an *N*-quaternized guanininium nucleobase. This hypothesis was supported by the H-8/C-6 <sup>4</sup>J<sub>CH</sub> correlation and the downfield chemical shift of C-4, and comparisons of <sup>13</sup>C shifts of **1** with published data for similar purine bases, e.g., herbipoline (7,9-dimethyl-2-(*N*-amino)guaninium) [25,27–30] with the same C-2″–*N*-10 bond. The NMR data are in good agreement with other natural alkylpuriniums: 7,9-dimethyl-2-(*N*-methyl)guaninium chloride [25] and *N*,*N*-dimethyl-1,3-dimethylherbipoline [30].

Although **1** is chiral, the weak optical activity ( $[\alpha]_{23}^D$  +2 (*c* 0.4, MeOH)) suggests a near-racemic mixture.

No.	$\delta_{ m H}$ a	$\delta_{\mathrm{H}}$ <sup>b</sup>	$\delta_{\rm C}{}^{\rm a}$	$\delta_{\rm N}{}^{\rm b}$	<sup>1</sup> H- <sup>13</sup> C HMBC <sup>a</sup>	<sup>1</sup> H- <sup>15</sup> N HMBC <sup>b</sup>
N-1′		10.60 (1H, d, 2.0)		131.3		
2'	7.57 (1H, s)	7.53 (1H, d, 2.0)	125.9		C-1", 3', 5', 3a', 7a'	
3′			113.4			
3a′			125.2			
4'	7.65 (1H, d, 8.5)	7.12 (1H, d, 8.5)	120.8		C-3′, 6′, 3a′, 7a′	
5'	7.24 (1H, dd, 8.5, 1.6)	7.45 (1H, dd, 8.5, 1.6)	124.2		C-7′, 3a′	
6'			117.0			
7′	7.60 (1H, d, 1.5)	7.61 (1H, d, 1.5)	115.9		C-6′, 5′, 3a′	
7a′	-	-	139.1			
N-1″				47.9 <sup>c</sup>		
2″	4.07 (1H, dd, 13.7, 5.9)4.15 (1H, dd, J = 13.7, 6.8 Hz)	4.07 (2H, m)	69.3		C-2", 3', -N(CH <sub>3</sub> ) <sub>3</sub>	N-10
3″	6.05 (1H, t, 6.3)	6.05 (1H, t, 6.3)	45.4		C-1", 3', 3a', 2', N-10	
1"-NMe	3.30 (9H, s)	3.27 (9H, s)	54.8		C-2", 1"	N-3″
N-1						
2			154.9			
N-3						
4			151.0			
5			110.0			
6			155.0			
N-7				156.4 <sup>c</sup>		
8	9.01 (1H, s)	8.81 (1H, s)	140.1		C-11, 4, 5	N-7, 9
N-9				157.3 <sup>c</sup>		
N-10				96.3 <sup>c</sup>		
11	4.11 (3H, s)	3.84 (3H, s)	36.2		C-5, 8	N-7
12	3.91 (3H, s)	3.92 (3H, s)	32.1		C-4, 8	N-9

Table 1. <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (150 MHz) spectroscopic data for geobarrettin D (1).

<sup>a</sup> Recorded in CD<sub>3</sub>OD. <sup>b</sup> Recorded in D<sub>2</sub>O/H<sub>2</sub>O (1/9). <sup>c 15</sup>N  $\delta$  obtained by indirect detection from <sup>1</sup>H-<sup>15</sup>N HMBC cross peaks.

# 2.2. Anti-Inflammatory Activity

When DCs were matured and activated in the presence of geobarrettin D (1), secretion of the pro-inflammatory cytokine IL-12p40 was diminished by 48%, whereas secretion of the anti-inflammatory cytokine IL-10 was not affected (Figure 3). On balance, these results indicate an overall anti-inflammatory effect of geobarrettin D (1).



**Figure 3.** The effect of geobarrettin D (1) on DC secretion of IL-12p40 and IL-10. DCs were matured and activated with IL-1 $\beta$ , TNF- $\alpha$  and LPS for 24 h in the absence (solvent control (CT)) or presence of geobarrettin D (1) at 10 µg/mL. The concentrations of IL-12p40 and IL-10 in the supernatants were determined by ELISA. The data are presented as SI, i.e., the concentration of each cytokine in the supernatant of cells cultured in the presence of compound 1 divided by the concentration of the cytokine in the supernatant of cells cultured without compound 1. The results are shown as mean  $\pm$  SEM, *n* = 3. Different from CT: \* *p* < 0.05.

### 3. Discussion

Marine sponges have proven to be a rich source of naturally occurring modified nucleosides: these exist as free bases, nucleotides, and within polynucleotides [31,32]. The first natural purinium salt found in nature, herbipoline, was isolated from the sponge *Geodia gigas* [32]. Subsequently, several related herbipoline salts were characterized from

tropical marine sponges: l-methylherbipoline from *Jaspis* sp. [25]], 1-methylherbipoline salts of halisulfate-1 and suvanine from *Coscinoderma mathewsi* [27] and suvanine (*N*,*N*-dimethyl-1,3-dimethylherbipoline salt) from *Coscinoderma* sp. [30]. A literature survey revealed that **1** is the first herbipoline-containing indole [32], and a rare bis-quaternized alkaloid.

Most natural product alkaloids are derived from the aromatic amino acids including tryptophan, tyrosine and phenylalanine. Compound **1** is likely biosynthetically derived from 6-bromotryptamine—an alkaloid known from other marine invertebrates [33] through a fusion of an oxidized tryptamine intermediate and a guanine equivalent (Figure 4) through a 1,4-conjugate addition, followed by extensive methylation reactions involving *S*-adenosylmethionine (SAM). Oxidation and conjugate addition is necessary and sufficient to explain formation of many C–C and C–heteroatom bonds in alkaloids [34], including **1**. For example, one of us (T.F.M) recently reported two cyclic guanidines, aiolochroiamides A and B, whose formation may also be rationalized by an oxidation–conjugate addition mechanism [35]. It is likely that the oxidation reaction is enzyme-mediated as spontaneous autoxidation seems unlikely, however the subsequent conjugate addition may be spontaneous given the low specific rotation (and therefore enantiomeric excess) observed for **1**. As with other complex highly-methylated quaternized alkaloids, the ordering of *N*-methylation and condensation reactions is uncertain. Further biosynthetic studies are necessary for understanding the biosynthesis of **1**, but these are beyond the scope of this study.



Figure 4. Putative biosynthesis of geobarrettin D (1).

Purines have found antiviral, antibiotic, and anticancer activities [27,31,36,37], and have the potential to regulate myocardial oxygen supply and cardiac blood flow [38]. In addition, evidence supports their role in biological evolution, differentiation, and ecological processes [31]. Purines are also involved in various inflammatory responses which underscores the significant attention given to purine natural products and their synthetic mimetics for the development of anti-inflammatory agents [20,39,40].

The anti-inflammatory properties of compound **1** were investigated in an in vitro model of human monocyte-derived DCs [20]. DCs matured and activated in the presence of compound **1** secreted less IL-12p40 than DCs cultured without **1**, whereas **1** had no effect on their secretion of IL-10. The pro-inflammatory cytokine IL-12p40 (one of the two chains that form the structures of IL-12 and IL-23 cytokines) is a major determinant of the differentiation of naïve T cells into Th1 or Th17 phenotypes [41], whereas the anti-inflammatory cytokine IL-10 drives polarization of naïve T cells into a T regulatory phenotype [42]. Thus, suppression of IL-12p40 secretion by DCs in the presence of **1** indicates that geobarrettin D (**1**) has anti-inflammatory activity.

#### 4. Materials and Methods

# 4.1. General Procedures

The UV spectrum was recorded on a NanoVueTM spectrophotometer (GE Healthcare Life Sciences, Little Chalfont, UK) with a 0.2 mm path length. Optical rotation was measured on a P-2000 polarimeter (Jasco, Oklahoma City, OK, USA), with a quartz cell (10 mm path length). The infrared spectrum was measured on a Spectrum Two TM FTIR spectrometer (Perkin Elmer<sup>®</sup>, Waltham, MA, USA) of samples as thin films. NMR spectra were recorded on a Bruker Avance 600 spectrometer (Billerica, MA, USA) (<sup>1</sup>H and  $^{13}$ C frequencies: 600.13 MHz and 150.76 MHz, respectively) in CD<sub>3</sub>OD and D<sub>2</sub>O/H<sub>2</sub>O (1:9). The residual solvent signals were used as internal references:  $\delta_{\rm H} 3.30 / \delta_{\rm C} 49.0$  ppm (CD<sub>3</sub>OD) and  $\delta_{\rm H}$  4.79 (D<sub>2</sub>O). For <sup>1</sup>H-<sup>15</sup>N 2D NMR spectroscopy, the nominal <sup>15</sup>N standard was liquid ammonia, NH<sub>3</sub> (l) ( $\delta$  = 0 ppm). Samples (1.6–6.0 mg) were introduced into Shigemi tubes and their 2D NMR spectra measured as illustrated by the following <sup>1</sup>H{<sup>13</sup>C} heteronuclear HSQC and HMBC spectra using modifications of the Bruker pulse sequences hsqcedetgpsisp2.4 and hmbcgplpndqf, respectively: Spectra were acquired at near ambient temperature (T = 300.0 K) in the specified solvent with an rf pulse calibrated to 1H  $\pi/2 = 8.75 \,\mu$ s, with appropriate gradient field strengths, and dwell times corresponding to  ${}^{1}$ H (F2) and  ${}^{13}$ C (F1) spectral widths of 8417.5 Hz and 33112.6 Hz and centered at  $\delta_{\rm H}$  7.01 and  $\delta_{\rm C}$  110.4 ppm, respectively. Accumulated scans (*n* = 8 for HMBC and *n* = 32 for HSQC) for each T1 increment were averaged between a relaxation delay, D1 = 1.5 s. The acquired matrix (<sup>1</sup>H and <sup>13</sup>C, 1024  $\times$  256 increments for HMBC, and 672  $\times$  256 for HSQC) was zero-filled in each dimension to a final size of  $2048 \times 1024$ , and processed, after standard apodizations, by Fourier transform. The high-resolution mass spectrum was measured on an Acquity UPLC I-Class System coupled to Xevo G2-XS QTof Mass Spectrometer (Waters<sup>®</sup>, Milford, MA, USA) using Acquity UPLC<sup>®</sup> HSS T3 column (High Strength Silica C18, 1.8  $\mu$ m, 2.1  $\times$  100 mm, Waters<sup>®</sup>, (Milford, MA, USA) operating at 60 °C). The MS and MS<sup>n</sup> spectra were recorded in positive mode and data were acquired using MassLynx<sup>®</sup> Software (version 4.1, Waters Crop., Milford, MA, USA). VLC chromatography on C18 adsorbent (LiChroprep RP-18, 40-63 µm, Merck Inc., Darmstadt, Germany) and Dionex 3000 HPLC system armed with a G1310A isopump, a G1322A degasser, a G1314A VWD detector (210 nm), a 250  $\times$  21.2 mm Phenomenex Luna C18(2) column (5  $\mu$ m), and a  $250 \times 4.6$  mm Phenomenex Gemini-NX C18 column (5  $\mu$ m) were conducted for separation and purification of pure compounds.

#### 4.2. Animal Materials

In short, the sponge material *Geodia barretti* was collected in Iceland, identified by. Hans Tore Rapp, University of Bergen (Norway), and deposited at University of Iceland. For a complete description of the samples, see Xiaxia Di et al. [20].

#### 4.3. Extraction and Isolation

Frozen sponge was cut into approximately 1 cm<sup>3</sup> pieces and freeze-dried. The dried tissue was extracted with a mixture of CH<sub>2</sub>Cl<sub>2</sub>:CH<sub>3</sub>OH (v/v, 1:1) for 3 times (2 L, each for 24 h) at room temperature. The combined CH<sub>2</sub>Cl<sub>2</sub>-MeOH extracts were dried under reduced pressure then the residue (1.8 g) was suspended in MeOH:H<sub>2</sub>O (v/v, 9:1) and subjected to a modified Kupchan partition, as previously described [21,22], to yield five fractions: hexane (fraction A), chloroform (fraction B), dichloromethane (fraction C), *n*-butanol (fraction D), and H<sub>2</sub>O (fraction E). Using a VLC RP-18 CC (MeOH-H<sub>2</sub>O, 10:90 $\rightarrow$ 100:0) technique, the mixture of the fractions B and C was separated into nice fractions (F2.1–F2.9). Fraction F2.2 (75.0 mg) was purified by preparative HPLC (28:72:0.1 CH<sub>3</sub>CN-H<sub>2</sub>O-TFA, 8.0 mL/min) and then re-chromatographed by semi-preparative HPLC (CH<sub>3</sub>CN-H<sub>2</sub>O-TFA, 31:69:0.1) to give geobarrettin D (1) (3.3 mg).

Geobarrettin D (1): Light yellowish oil;  $[\alpha]_{23}^D$  +2 (*c* 0.4, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) nm: 212 (3.90), 228 (4.04), 261 (3.81), 287 (3.65); IR  $\nu_{max}$  cm<sup>-1</sup>: 3255, 1679, 1607, 1478, 1444, 1385, 1203, 1131; For <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; HRESIMS *m*/*z* 458.1343 [M]<sup>+</sup> (C<sub>20</sub>H<sub>25</sub>ON<sub>7</sub>Br, 458.1298).

## 4.4. Maturation and Activation of DCs

DCs were differentiated from human monocytes as previously described [20]. Peripheral blood mononuclear cells (PBMCs) were isolated from peripheral blood obtained from healthy human donors (approval #06-068-V1 by National Bioethics Committee of Iceland (Visindasidanefnd) from 15<sup>th</sup> of December 2015) by density centrifugation

using Histopaque-1077 (Sigma-Aldrich, Munich, Germany). Then CD14<sup>+</sup> monocytes were isolated from the PBMCs using magnetic cell sorting and CD14 Microbeads (Miltenyi Biotech, Bergisch Gladbach, Germany). The CD14<sup>+</sup> monocytes were cultured at  $5 \times 10^5$  cells/mL in RPMI 1640 medium, supplemented with 10% fetal calf serum and 5% penicillin/streptomycin (all from Gibco<sup>®</sup>, Thermo Fisher Scientific, UK) in 48 well tissue culture plates for seven days. In order to differentiate CD14<sup>+</sup> monocytes into immature DCs, IL-4 at 12.5 ng/mL and GM-CSF at 25 ng/mL (both from R&D Systems, Bio-Techne, Abingdon, England) were added to the cells. After seven days the monocytes had differentiated into immature DCs which were harvested and cultured for 24 h in 48 well tissue culture plates at  $2.5 \times 10^5$  cells/mL. The immature DCs were matured and activated by culturing them with IL-1 $\beta$  at 10 ng/mL, TNF- $\alpha$  at 50 ng/mL (both from R&D Systems), and lipopolysaccharide (LPS) at 500 ng/mL (Sigma-Aldrich, Munich, Germany). Geobarrettin D (1) was dissolved in DMSO and added to the DCs at  $10 \,\mu g/mL$  at the same time as the cytokines and LPS. DMSO was used as a control. After 24 h the mature and activated DCs were harvested and the concentrations of IL-12p40 and IL-10 in the supernatants were measured by sandwich ELISA using DuoSets from R&D Systems according to the manufacturer's protocol. The results are expressed as a secretion index (SI). Data are presented as the mean values  $\pm$  SEM, n = 3. As the data were not normally distributed, Mann–Whitney U test was used to determine statistical differences between the groups (SigmaStat 3.1, Systat Software, San Jose, CA, USA) and p < 0.05 was considered as statistically significant.

## 5. Conclusions

Geobarrettin D (1) is a newly detected bromoindole alkaloid possessing an unusual brominated and fused-herbipoline-dimethylguaninium heterocycle. Geobarrettin D (1) decreased DC secretion of the pro-inflammatory cytokine IL-12p40, indicating its potential as an anti-inflammatory agent.

**Supplementary Materials:** The following are available online at https://www.mdpi.com/article/10 .3390/molecules28072937/s1, Figure S1. 1H NMR spectrum of geobarrettin D (1), recorded in CD3OD, 600 MHz; Figure S2. 13C NMR spectrum of geobarrettin D (1), recorded in CD3OD, 150 MHz; Figure S3. DEPT-135 NMR spectrum of geobarrettin D (1); Figure S4. HSQC spectrum of geobarrettin D (1); Figure S5. HMBC spectrum of geobarrettin D (1); Figure S6. COSY spectrum of geobarrettin D (1); Figure S7. 1H NMR spectrum of geobarrettin D (1), recorded in D2O/H2O 10/90, 600 MHz; Figure S8. 1H-15N HSQC spectrum of geobarrettin D (1), recorded in D2O/H2O 10/90; Figure S9. 1H-15N HMBC spectrum of geobarrettin D (1), recorded in D2O/H2O 10/90; Figure S9. 1H-15N HMBC spectrum of geobarrettin D (1), recorded in D2O/H2O 10/90; Figure S10. ESI spectrum of geobarrettin D (1); Figure S11. IR spectrum of geobarrettin D (1).

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